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"Telomerisation of trifluoroethene with acetone"

"2-Deoxy-2-fluoro-1,3,5-tri-O-benzoyl-a-D-arabinofuranose (3)"

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"... gave needles: mp 83–85 °C; $[\alpha]_D^{25}$ –110° (*c* 1.4, CHCl₃); IR (KBr); *v* 1730(s) and 1260 (ester), 860 and 840 (Me₃Si), and 710(m) cm⁻¹ (Ph); ¹H NMR ..."

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¹⁵F NMR (56.4 MHz, CC1₃F): 6-88.0 (m, 1F, NF), -68.3 (dt, 3F, J = 13.5 Hz, CF₃N), -82.0 (tt, 3F, J = 13.5 Hz, CF₂CF₂CF₂CF₃).

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[1] V.A. Petrov, C.G. Krespan, B.E. Smart, J. Fluorine Chem. 77 (1996) 139–142.

Patent:

[2] E.T. McBee, O.R. Pierce, H.M. Metz, US Patent 2 899 454 (1959).

Abstract of paper presented at meeting:

[3] J.G. Riess, in: Proceedings of the 219th American Chemical Society meeting on blood substitutes, liquid breathing, and more, all with fluorocarbons, San Francisco, CA, 27 March 2000, Fluorine Division, Paper 17. *Book:*

[4] W.A. Sheppard, C.M. Sharts, Organic Fluorine Chemistry, W.A. Benjamin, New York, 1997, pp. 39–96. *An article in an edited book:*

[5] H.J. Eméleus, Metallic compounds containing fluorocarbon radicals and organometallic compounds containing fluorine, in: J.H. Simons (Ed.), Fluorine Chemistry, Vol. 2, Academic Press, New York, 1954, pp. 321–336 (Footnote). **Tables**: Many types of data can be most effectively presented in tabular form, authors should consult a current issue for examples. Tables are numbered in sequence as they are to appear in the text, using **Arabic** numerals. *Vertical rules should not be used*. Tables should be provided for reproduction at single or double column width. *Title*: The title should accurately and concisely describe the content of the table. Each column should have a *Column heading*: these should be short – use abbreviations liberally. Most important, column headings should name the parameter and unit of measure listed in the column. *Footnotes*: to a table may be used to expand column headings and are indicated by superscript, lower case letters^{a, b, c}.

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$$\frac{spacer n R}{10 - N = CH - 1 CN}$$

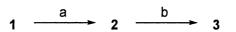
$$\frac{11}{11} - NHC(O) - 2 OMe$$

$$\frac{12}{12} - C(O)O - 4 F$$

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$$1 \xrightarrow{\text{LiN}_3, \text{THF}} 2 \xrightarrow{\text{H}_2, \text{Pd-C}} 3$$

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